

Application Leaflet

BENTONE® Hectorite based grades

Unique mineral for
manifold applications

Unique chemistry, sustainable solutions



Key Benefits

- Effective viscosity build in aqueous and non-aqueous applications
- Significantly higher efficiency than other mineral based rheology modifiers such as bentonite, Sepiolite and attapulgite
- Strong adsorptive effect due to large surface area or applications in water treatment

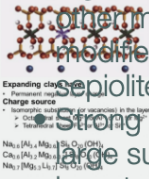


FIGURE 1: Mineral structure of Hectorite

Introduction

Inorganic rheology modifiers based on hectorite and bentonite clays are widely used in the formulation of manifold aqueous and, with specific modification, also for non-aqueous systems to provide excellent stability and anti-sag stability combined with efficient low-shear viscosity build as well as thixotropic flow characteristic.

Mineralogy

Clay minerals are abundant in nature and are essentially composed of some of the most common elements found in the Earth's crust. Clay minerals are crystalline, layered hydrous aluminum- or magnesium- silicates and therefore have two-dimensional building blocks. Hence, a clay layer is a platelet of nanoscopic thickness, which stacks to form a three dimensional solid material. Clays can be classified based upon the composition of the individual clay platelet which typically contains either one or two silicate sheets condensed on a (central) magnesium or aluminum hydroxide sheet. The group of 2:1 clay platelet, to which smectite clays such as Hectorite and Bentonite belong to, contain two silica sheets and one central magnesium hydroxide or aluminum hydroxide sheet. smectite clays experience charge deficiency occurring in the crystalline structure when a number of metal cations take up positions that are normally occupied by cations of higher valency. These so called isomorphous substitutions leave a net negative charge on the clay platelet which is neutralized with exchangeable cations positioned on the surfaces of the clay platelets.

FIGURE 1 provides an overview on the mineral structure of smectite clays.

Hectorite clay is a natural occurring clay that belongs to the group of smectite clay. It is obtained when some magnesium cations are substituted by lithium in the clay layer structure. The negative charge on the platelet resides within the central octahedral sheet of the clay structure and is countered by exchangeable sodium to maintain magnetic neutrality.

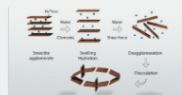


FIGURE 2: Mechanism of activation

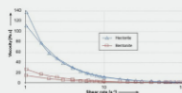


FIGURE 3: Comparison of efficiency

Additionally to differences in the composition of the clay compared to other members of the smectite group, size and shape of the individual platelets play a major role in its performance. The size of Hectorite is approximately 10 times smaller than Bentonite. As the thickening effect is driven by the edge length of the platelets, the resulting longer edge length of the hectorite offers a major benefit after activation.

Activation and mechanism

Hectorite clay can be activated by delamination of the platelets driven by swelling in pure water with high shear dispersion (FIGURE 2). The sodium ions hydrate, pushing the platelets apart and the clay swells. The high shear forces during dispersion cause the platelets to move away from each other as colloidal, slightly charged particles. After removal of shear, double-layer effects, van der Waals forces of attraction and electrostatic repulsion result in the flocculation edge-to-edge, which is visible as the desired viscosity is formed.

Due to the longer edge length of hectorite in comparison to bentonite, it builds a much higher low shear viscosity when formulated at an equal concentration in water (FIGURE 3). The thixotropic effect is also significantly stronger. Hectorite however require higher shear during delamination due to its smaller particle size.

Hectorite clays are being formulated in various systems. Besides the mentioned application in the market segment of paints and coatings, hectorite clays find use in ceramics laundry, cosmetics, medicine, pottery, porcelain etc. A further famous application is water treatment due to the strong adsorption power of the platelet surfaces.

PICTURE 1



In addition, the platelet surfaces can be modified in a reaction with quaternary ammonium compounds converting the clay into a hydrophobic form that is used for solvent-borne systems – an organoclay. Depending on the type of quaternary ammonium species used, products can be designed for all different polarities and chemistries of solvents. Hectorite based organoclays dominate in performance in comparison to similar Bentonite based grades.

These products are commercially available under the registered BENTONE® and BENAQUA® name.

BENTONE® Hectorite grades

Aqueous Application	Non-aqueous application
BENTONE® EW	BENTONE® 27
BENTONE® LT	BENTONE® 38
BENTONE® HC	BENTONE® SD-1
BENTONE® DE	BENTONE® P 270 CO
BENTONE® DH-CE*	BENTONE® P 380 MS
BENTONE® CT	
BENTONE® OC	
BENAQUA® 4000	

* available only in the USA

NOTE:

The information herein is currently believed to be accurate. We do not guarantee its accuracy. Purchasers shall not rely on statements herein when purchasing any products. Purchasers should make their own investigations to determine if such products are suitable for a particular use. The products discussed are sold without warranty, express or implied, including a warranty of merchantability and fitness for use. Purchasers will be subject to a separate agreement which will not incorporate this document.

© Copyright 2024, Elementis Specialties, Inc. All rights reserved. Copying and/or downloading of this document or information therein for republication is not allowed unless prior written agreement is obtained from Elementis Specialties, Inc.

August 2024

For more details
please contact:

North America

Elementis
469 Old Trenton Road
East Windsor,
NJ 08512, USA
Tel: +1 609 443 2500

Europe

Elementis UK Ltd c/o
Porto Business Plaza
Santos Pousada Street, 290
4300-189, Porto, Portugal

Asia

Deuchem (Shanghai) Chemical
Co., Ltd.
99, Lianyang Road
Songjiang Industrial Zone
Shanghai, China 201613
Tel: +86 21 577 40348

elementis.com



Unique chemistry,
sustainable solutions